

(1.67×10^5)
 (27)
 (29)
 $(3D)$
 (1121)
 (746)
 (500)
 (1356)
 (0.035%)
 $(\approx 100\%)$

2. Experimental

2.1. Preparation of CC@ZIF-67

$(90 \text{ C} \cdot 6)$
 (100 L)
 $(2\text{-MI}, 1.64)$
 (60 C)

2.2. Preparation of CC@Co₃O₄-PCNA and CC@Co₄N-PCNA

$(600 \text{ C} \cdot 2 \text{ C}^{-1})$
 $(330 \text{ C} \cdot 2)$
 $(600 \text{ C} \cdot 2 \text{ C}^{-1})$
 (50)
 $(\text{CC}@ \text{IF-67})$
 $(\text{CC}@ \text{C}_3\text{A})$
 $(\text{CC}@ \text{C}_4 - \text{C A})$

2.3. Preparation of CC@PCNA, CC@Co₄N NWs, ZIF-67 and Co₄N-PC

$(\text{CC}@ \text{IF-67})$
 $(2 \text{ C}^{-1} \cdot 2)$
 (1M)
 $(\text{CC}@ \text{C}_3\text{A})$
 $(\text{CC}@ \text{C}_3\text{A})$
 $(\text{CC}@ \text{C}_4 - \text{C A})$
 $(\text{CC}@ \text{C}_3\text{A})$
 $(\text{CC}@ \text{C}_4 - \text{C A})$
 $(\text{CC}@ \text{C}_4 - \text{C A})$

(1.64g)
 (50 L)
 (70 C)
 (IF-67)
 $(\text{CC}@ \text{C}_4 - \text{C A})$

2.4. Preparation of the sulfur composites

$(\text{CC}@ \text{C}_4 - \text{C A})$
 $(\text{CC}@ \text{C}_3\text{A} - \text{C A})$
 $(\text{CC}@ \text{C}_4 - \text{C A})$
 (0.15)
 (10 L)
 $(50 \text{ C} \cdot 6)$
 $(155 \text{ C} \cdot 12)$
 $(1.01\text{--}6.20)$

2.5. Preparation of the Li₂S₆ solution

(L_2)
 $(1:5)$
 (HF)

2.6. Material characterization

(FE-EM)
 (EM)
 (D/MA)
 (IC)
 (GA)
 (A L200B)

2.7. Electrochemical measurement

(F)
 (2025)
 (0.1)
 (1.0 M L F I)
 (D L/DME)
 (D L)
 (C 2001A)
 (EI)
 (M -300)

2.8. Density functional theory calculations

(DF)
 (A)
 $(51, 52)$
 (53)
 (54)
 (A)
 (F)
 (M)
 (55)

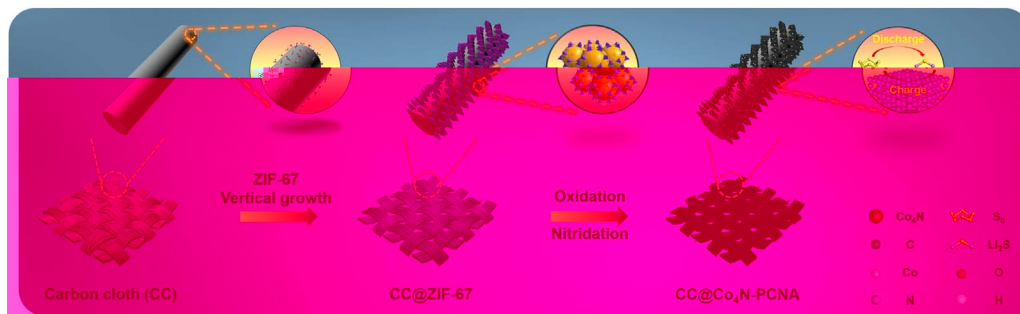


Fig. 1. Schematic diagram of the synthesis process for CC@Co₃N-PCNA.

$$E_b = E_{Li_2S_n} + E_{Co_4N} - E_{Li_2S_n + Co_4N} \quad (1)$$

3. Results and discussion

Figure 1 shows the synthesis process for CC@Co₃N-PCNA. The process starts with carbon cloth (CC) and ZIF-67. ZIF-67 is grown vertically on CC to form CC@ZIF-67. This intermediate is then subjected to oxidation and nitridation to form CC@Co₃N-PCNA. The legend identifies the components: Co₃N (red circle), C (black circle), Co (orange circle), N (blue circle), S₈ (yellow star), Li₂S (red line), O (red circle), and H (pink circle).

Figure 2 shows the FE-EM images of (a) IF-67, (b) CC@IF-67, (c) CC@C₃N-PCNA, and (d) CC@C₄-CA. The images show the morphology of the materials, with IF-67 appearing as small particles, CC@IF-67 as a network of interconnected particles, and CC@C₃N-PCNA and CC@C₄-CA as larger, more complex structures. The scale bars indicate the size of the particles in micrometers.

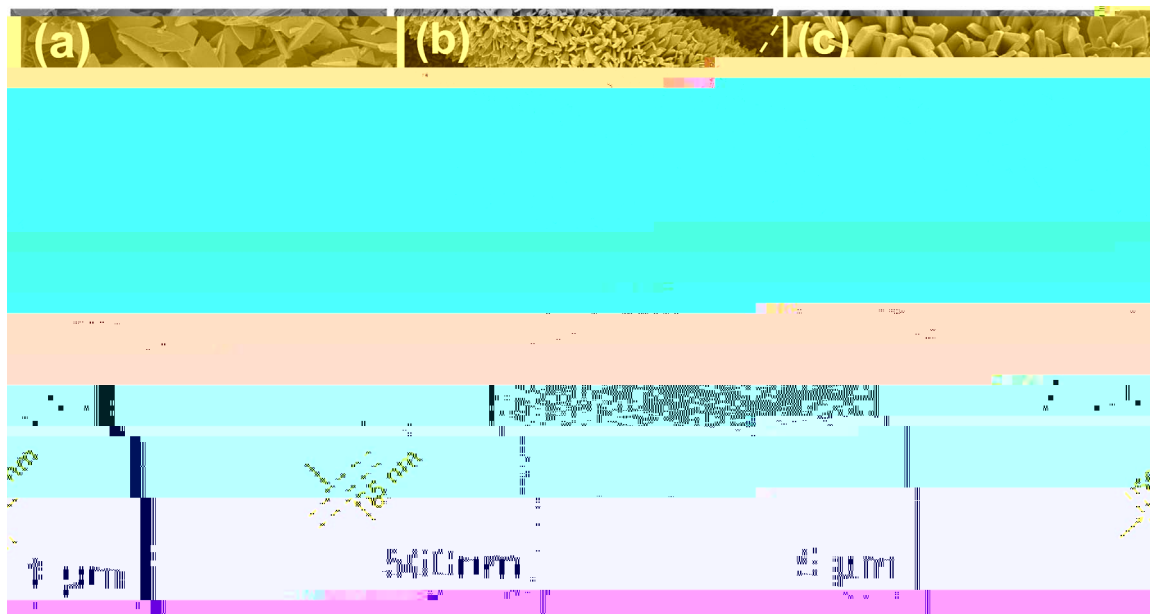


Fig. 2. FE-EM images of (a) IF-67, (b) CC@IF-67, (c) CC@C₃N-PCNA, and (d) CC@C₄-CA.

CC@ IF-67 . CC@C₄ - C A (EM).

CC@ IF-67 (F . 3). A . 1.5 μ

(F . 3 . -). F . fi . EM . F . 3 .

. 40

H . EM (H - EM)

(v F . 3 F . 5 , (111) C₄

(311) C_{3 4} , C_{3 4} (Fd3m (227))

C₄ (Pm3m (221)) . (F . 5 . F . 3)

C₄ C - C

30 .

D CC, CC@C₄ - C A . CC@C_{3 4} - C A

(v F . 4 . ff . 25.6

(002) C_{3 4} (DF#42-1467) - fi

(v 0.243 , (311)

C_{3 4} (F . 5) . CC@C₄ - C A , (

600 C₃ (v H₃ - fi v 50

D (F . 6) (v C -

ff .

C_{3 4} C₄ (v 31 :

24Co₃O₄+64NH₃ = 18Co₄N + 23N₂+96H₂O (2)

A (v F . 4 , C₄

(DF#41-0943) (v (111) , (200) (220) , (v

ff, “ff” L
 C₄ (w
 L_{2,2}/L₂
 ff
 I
 /CC@C₄ - C A (w
 285⁻¹ 660⁻¹, (C
 5), (F, 21,
 L, B

4. Conclusions

I, (w C₄ -
 L, B.
 (w C₄ - ff L ff
 ff B ff
 L CC@
 C₄ - C A, /CC@C₄ - C A (w
 1121 A⁻¹ 0.5 C₄ 100
 746 A⁻¹ 5 C₄ (w
 0.035% 500 5 C₄ (w (w
 fl

Acknowledgment

(w F
 C (G. 51672033, 1610105, 1610255), F
 K L I E A
 M, M E, C (KLI EAM 201601),
 F L E, D
 , C (2016-54).

Appendix A. Supporting information

(w
 :10.1016/. 2018.12.005

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